

C2—O3—C3	60.7 (2)	C8—C9—C4	120.5 (3)
C10—O5—P1	123.3 (2)	O2—C12—N1	127.9 (2)
C11—O6—P1	121.1 (2)	O2—C12—O1	123.0 (2)
C12—N1—C13	125.4 (2)	N1—C12—O1	109.1 (2)
O1—C1—C2	109.4 (2)	C14—C13—C18	119.6 (3)
O1—C1—P1	108.0 (2)	C14—C13—N1	117.2 (2)
C2—C1—P1	110.9 (2)	C18—C13—N1	123.2 (2)
O3—C2—C3	60.5 (2)	C13—C14—C15	120.2 (3)
O3—C2—C1	115.4 (2)	C16—C15—C14	120.5 (3)
C3—C2—C1	121.1 (2)	C17—C16—C15	118.9 (3)
O3—C3—C2	58.80 (15)	C16—C17—C18	121.5 (3)
O3—C3—C4	118.1 (2)	C17—C18—C13	119.3 (3)
C2—C3—C4	124.1 (2)		

Intensity decay was monitored by recollecting the first 50 frames at the end of data collection (8.5 h). A total of 1325 frames (15 s frame⁻¹, 0.3° scan width) was collected. The structure was solved by direct methods and refined successfully in the monoclinic space group *P2₁/c*. Full-matrix least-squares refinement was carried out by minimizing $w(F_o^2 - F_c^2)^2$. The non-H atoms were refined anisotropically, whereas H atoms connected to C1, C2, C3 and N1 were refined isotropically to convergence. The rest of the H atoms were refined using an appropriate riding model. Very fine needles of 0.15 mm in the largest dimension were obtained initially from the diethyl ether solution and data collection was attempted on these crystals. To our surprise, the structure could be solved and partially refined using the CCD system. We are convinced that with the use of sensitive detectors such as the CCD area-detector system, single-crystal X-ray diffraction can truly be used as an analytical tool. Some pertinent data for the small crystal are presented here: $a = 11.9099$ (14), $b = 9.4878$ (11), $c = 17.946$ (2) Å, $\beta = 108.660$ (3)°, $V = 1921.2$ (4) Å³; 6367 reflections collected, 1793 independent reflections, $R1 = 0.103$, $wR2 = 0.260$, crystal dimensions = $0.15 \times <0.03 \times <0.03$ mm, data collection time 21 h (45 s frame⁻¹).

Data collection: *SMART* (Siemens, 1995). Cell refinement: *SAINT* (Siemens, 1995). Data reduction: *SAINT*. Program(s) used to solve structure: *SHELXS86* (Sheldrick, 1990). Program(s) used to refine structure: *SHELXL93* (Sheldrick, 1993). Molecular graphics: *SHELXTL-Plus* (Sheldrick, 1995). Software used to prepare material for publication: *SHELXTL-Plus*.

We are grateful to the University of Missouri–St. Louis RIA Fund for support of this project and the National Science Foundation for a grant to purchase the X-ray diffractometer (CHE-9309690).

Supplementary data for this paper are available from the IUCr electronic archives (Reference: PT1044). Services for accessing these data are described at the back of the journal.

References

- Al-Ajlouni, A. M. & Espenson, J. H. (1995). *J. Am. Chem. Soc.* **117**, 9243.
 Boehlow, T. R. & Spilling, C. D. (1997). *Tetrahedron Lett.* Submitted.
 Heaney, H. (1993). *Aldrichchim. Acta*, **26**, 35.
 Herrmann, W. A., Fischer, R. W. & Marz, D. (1991). *Angew. Chem. Int. Ed. Engl.* **30**, 1638.
 Herrmann, W. A., Fischer, R. W., Rauch, M. U. & Scherer, W. (1994). *J. Mol. Catal.* **86**, 243.
 Herrmann, W. A., Fischer, R. W., Scherer, W. & Wauch, M. U. (1993). *Angew. Chem. Int. Ed. Engl.* **32**, 1157.

- Murray, R. W., Singh, M., Williams, B. L. & Moncrieff, H. M. (1995). *Tetrahedron Lett.* **36**, 2437.
 Sheldrick, G. M. (1990). *Acta Cryst.* **A46**, 467–473.
 Sheldrick, G. M. (1993). *SHELXL93. Program for the Refinement of Crystal Structures*. University of Göttingen, Germany.
 Sheldrick, G. M. (1995). *SHELXTL-Plus*. Release 5.03. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.
 Siemens (1995). *SMART and SAINT. Data Collection and Processing Software for the SMART System*. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.

Acta Cryst. (1997). **C53**, 1949–1950

The Khellin Quinone

ABDERRAHMAN YAKOUBI,^a JALEL EDDINE HILA,^b
 BOUGHZALA HABIB^c AND ABDELMALEK BIZID^a

^aLaboratoire A. Kallel de Cristallographie, Département de Physique, Faculté des Sciences, 1060 Campus Universitaire, Tunis, Tunisia, ^bSociété Arabe des Industries Pharmaceutiques, 126 Av. Jugurtha, 1002 Mutuelleville, Tunis, Tunisia, and ^cDépartement de Chimie, Faculté des Sciences, Campus Universitaire, 1060 Tunis, Tunisia. E-mail: kamel.benrhouma@cck.rnrt.tn

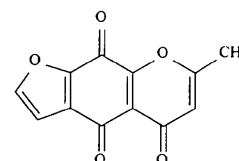
(Received 1 April 1997; accepted 14 July 1997)

Abstract

The title compound, 2-dehydro-2-methyl-5,8-dioxo-5,8-dihydrofuro[3,2-*g*]chromen-4-one, C₁₂H₆O₅, derives from the khellin molecule (7-methyl-4,9-dimethoxy-5H-furo[3,2-*g*][1]benzopyran-5-one). The molecular skeleton is nearly planar as in all furobenzopyranones. The intermolecular interactions are strengthened by C—H···O bonds.

Comment

The title compound, (I), has been studied in order to elucidate the transformation of khellin to khellinquinone.



(I)

The molecular skeleton is nearly planar (Fig. 1) as in all furobenzopyranones (El-Sayed, Ammon & Abd El-Rahman, 1988) and the crystal structure is made up of layers. The maximum deviation from the least-

squares plane through atoms C1–C12, O1, O3 and O5 is 0.108 (2) Å for C9. Similar to other furobenzopyranones, we have found that the O2···O4 distance of 2.808 (2) Å is close to the usual O···O contact distance of 2.8 Å (El-Sayed, Ammon & Abd El-Rahman, 1988). The crystal cohesion is strengthened by non linear C—H···O hydrogen-bonding interactions (Table 2).

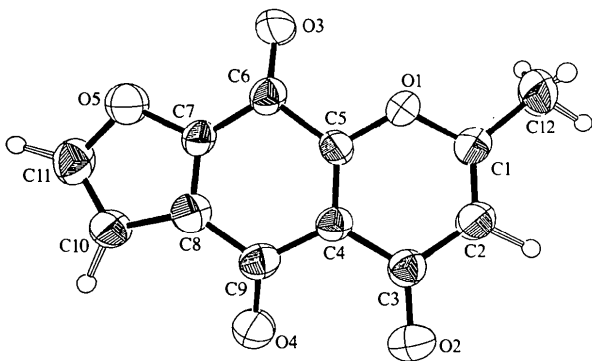


Fig. 1. ORTEP (Johnson, 1971) view of the molecule with the atomic numbering. Displacement ellipsoids are drawn at the 50% probability level. H atoms are shown as small spheres of arbitrary radii.

Experimental

The title compound was prepared by the method of Sreerama Murthi & Seshadri (1949). When khellin was treated with nitric acid, the khellin-quinone separated as an orange solid which was filtered and washed free of acid and crystallized from alcohol. Crystals were grown as described previously (Sreerama Murthi & Seshadri, 1949; Hila, 1983).

Crystal data

C₁₂H₆O₅
M_r = 230.17
 Monoclinic
*P*2₁/*c*
a = 7.552 (3) Å
b = 8.1720 (10) Å
c = 15.624 (3) Å
 β = 99.29 (2)°
V = 951.6 (4) Å³
Z = 4
D_x = 1.607 Mg m⁻³
D_m not measured

Data collection

Enraf–Nonius CAD-4 diffractometer
 $\omega/2\theta$ scans
 Absorption correction: none
 2154 measured reflections
 2076 independent reflections
 1353 reflections with *I* > 4 σ (*I*)

Mo K α radiation
 λ = 0.71069 Å
 Cell parameters from 25 reflections
 θ = 13–15°
 μ = 0.128 mm⁻¹
T = 293 (2) K
 Parallelepiped
 0.30 × 0.08 × 0.05 mm
 Orange

*R*_{int} = 0.010
 θ_{\max} = 26.97°
h = -9 → 9
k = -10 → 0
l = 0 → 19
 1 standard reflection
 frequency: 120 min
 intensity decay: 0.4%

Refinement

Refinement on *F*²
R[*F*² > 2 σ (*F*²)] = 0.046
 wR (*F*²) = 0.121
S = 1.093
 2076 reflections
 179 parameters
 All H atoms refined
 $w = 1/[\sigma^2(F_o^2) + (0.051P)^2 + 0.2658P]$
 where *P* = (*F_o*² + 2*F_c*²)/3

(Δ/σ)_{max} = -0.001
 $\Delta\rho_{\max}$ = 0.207 e Å⁻³
 $\Delta\rho_{\min}$ = -0.174 e Å⁻³
 Extinction correction: SHELXL93
 Extinction coefficient: 0.0041 (13)
 Scattering factors from International Tables for Crystallography (Vol. C)

Table 1. Selected geometric parameters (Å, °)

C1—O1	1.380 (2)	C6—O3	1.209 (2)
C1—C12	1.475 (3)	C7—O5	1.364 (2)
C3—O2	1.227 (2)	C9—O4	1.208 (2)
C4—C9	1.493 (3)	C11—O5	1.370 (3)
C5—O1	1.340 (2)		
C2—C1—C12	127.4 (2)	O3—C6—C5	121.9 (2)
O1—C1—C12	111.3 (2)	O4—C9—C8	121.0 (2)
O2—C3—C2	122.8 (2)	O4—C9—C4	123.4 (2)
O2—C3—C4	123.4 (2)	C5—O1—C1	118.9 (2)
O3—C6—C7	126.8 (2)	C7—O5—C11	105.3 (2)

Table 2. Hydrogen-bonding geometry (Å, °)

<i>D</i> —H··· <i>A</i>	<i>D</i> —H	H··· <i>A</i>	<i>D</i> ··· <i>A</i>	<i>D</i> —H··· <i>A</i>
C2—H2···O2 ⁱⁱⁱ	0.97 (2)	2.50 (2)	3.462 (3)	173 (3)
C11—H11···O3 ⁱⁱⁱ	0.95 (2)	2.47 (2)	3.389 (3)	161 (3)
C12—H12···O4 ⁱⁱⁱ	0.85 (3)	2.51 (3)	3.232 (3)	143 (3)
C11—H11···O5 ⁱⁱⁱ	0.95 (2)	2.65 (3)	3.275 (3)	123 (3)
C12—H12···O1 ^{iv}	1.01 (5)	2.67 (5)	3.644 (4)	162 (3)

Symmetry codes: (i) 2 - *x*, 1 - *y*, - *z*; (ii) - *x*, $\frac{1}{2}$ + *y*, $\frac{1}{2}$ - *z*; (iii) *x*, *y* - 1, *z*; (iv) 1 - *x*, - *y*, - *z*.

Data collection: CAD-4 EXPRESS (Enraf–Nonius, 1994). Cell refinement: CAD-4 EXPRESS. Data reduction: MolEN (Fair, 1990). Program(s) used to solve structure: MULTAN80 (Main *et al.*, 1980). Program(s) used to refine structure: SHELXL93 (Sheldrick, 1993). Molecular graphics: ORTEP (Johnson, 1971). Software used to prepare material for publication: SHELXL93.

Supplementary data for this paper are available from the IUCr electronic archives (Reference: AB1487). Services for accessing these data are described at the back of the journal.

References

- El-Sayed, K., Ammon, H. L. & Abd El-Rahman, A. M. (1988). *Acta Cryst.* **C44**, 676–678.
- Enraf–Nonius (1994). CAD-4 EXPRESS. Version 5.1/1.2. Enraf–Nonius, Delft, The Netherlands.
- Fair, C. K. (1990). *MolEN. An Interactive Intelligent System for Crystal Structure Analysis*. Enraf–Nonius, Delft, The Netherlands.
- Hila, J. E. (1983). Thesis, University of Paris XI, France.
- Johnson, C. K. (1971). ORTEP. Report ORNL-3794, revised. Oak Ridge National Laboratory, Tennessee, USA.
- Main, P., Fiske, S. J., Hull, S. E., Lessinger, L., Germain, G., Declercq, J.-P. & Woolfson, M. M. (1980). *MULTAN80. A System of Computer Programs for the Automatic Solution of Crystal Structures from X-ray Diffraction Data*. Universities of York, England, and Louvain, Belgium.
- Sheldrick, G. M. (1993). *SHELXL93. Program for the Refinement of Crystal Structures*. University of Göttingen, Germany.
- Sreerama Murthi, V. V. & Seshadri, T. R. (1949). *Proc. Indian Acad. Sci. Sect. A*, **30**, 107–113.